# **Prediction of the main macrocyclic conjugation pathway for porphyrinoids from the ring current distribution†**

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For some porphyrinoids, such as orangarin and amethyrin, the main route of macrocyclic  $\pi$ -circulation is different from the main macrocyclic conjugation pathway predicted by porphyrin chemists. Our analytical theory of ring-current diamagnetism allows us to predict the main macrocyclic conjugation pathway from the ring current distribution. We can now interpret macrocyclic aromaticity and macrocyclic circulation consistently within the same theoretical framework.

## **1. Introduction**

During the last two decades, many expanded, contracted, confused, and fused porphyrins have been synthesized.**1–6** Porphyrin chemists used to account for the electronic, magnetic, and chemical properties of porphyrins by a formal analogy between the main conjugation pathway in the macrocycle and the corresponding monocyclic annulene.<sup>1-6</sup> For example, natural porphyrins are described as bridged diaza[18]annulenes, and the same convention has been applied to many expanded porphyrins and porphyrinoids.**1–5** Porphyrinoids have often been named systematically after the annulene model.**<sup>2</sup>** In fact, not only macrocyclic conjugation but also pyrrole rings are major sources of global aromaticity.**5–7** However, it is generally true that macrocyclic conjugation is responsible primarily for the chemical shifts of protons attached to the porphyrinoid macrocycles.**1-6** To emphasize this fact, porphyrin chemists refer to macrocyclic aromaticity simply as aromaticity. PAPER<br>
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Macrocyclic aromaticity and macrocyclic circulation are two different manifestations of macrocyclic conjugation in porphyrinoids.**1-6** A ring current distribution in a polycyclic  $\pi$ -system, however, is strongly dependent on molecular geometry, so that information on global and macrocyclic aromaticity cannot be extracted directly from it. For some expanded porphyrins, such as orangarin and amethyrin, the main stream of  $\pi$ -circulation<sup>8,9</sup> deviates from the so-called main macrocyclic conjugation pathway**5,10** even if the direction of the macrocyclic ring current might be predicted from the annulene picture. In this paper, we propose a general theoretical method for extracting information on the main macrocyclic conjugation pathway from the ring current distribution, and interpret macrocyclic conjugation and macrocyclic circulation within the same theoretical framework.

## **2. Terminology**

Certain terms in porphyrin chemistry have sometimes been used to mean different things;**1-12** however, in this paper, the following meanings are applied. *Circuits*stand for all possible cyclic or closed paths that can be chosen from a cyclic  $\pi$ -system.<sup>13</sup> As shown in Fig. 1, 11 non-identical circuits and 20 circuits in all can be chosen from the  $\pi$ -system of porphine (1).<sup>7</sup> As in classical chemistry,<sup>14</sup> the term *aromatic* describes molecules that benefit energetically from the delocalization of  $\pi$ -electrons in closed circuits. *Topological resonance energy* (TRE), a kind of aromatic stabilization energy (ASE), is employed as a primary criterion of aromaticity,**11,12** which arises from a collection of circuits.



**Fig. 1** Non-identical circuits in porphine (**1**).

The term *macrocycle* is used to represent a planar or quasiplanar polycyclic  $\pi$ -system with a large inner cavity. Porphyrins and kekulene are such macrocyclic compounds. A *macrocyclic circuit* is any large circuit that surrounds the inner cavity (*e.g.*, circuits **c**–**k** in Fig. 1), also being referred to as a *conjugation pathway* in the macrocycle. This is distinguished from a *local*

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*circuit* located in each five-membered ring (*e.g.*, circuits **a** and **b** in Fig. 1). All macrocyclic circuits constitute *macrocyclic conjugation*. **5-7** *Macrocyclic aromaticity* and *macrocyclic circulation* are the aromaticity and  $\pi$ -circulation associated with macrocyclic circuits, respectively. *Superaromatic stabilization energy* (SSE) represents an extra ASE due to macrocyclic conjugation.**5-7,15,16** Thus, the term *superaromaticity* is a synonym of macrocyclic aromaticity.

#### **3. Theoretical background**

The concept of bond resonance energy (BRE) is useful for justifying the idea of macrocyclic aromaticity in porphyrin chemistry.**5-7** BRE is defined as follows.<sup>17-19</sup> A hypothetical  $\pi$ -system, in which a given  $\pi$ -bond (*e.g.*, a  $\pi$ -bond formed between the *p*th and *q*th atoms) interrupts the cyclic conjugation at that point, is constructed by multiplying  $\beta_{pq}$  by i and  $\beta_{qp}$  by -i, where  $\beta_{pq}$  is the resonance integral between the two conjugated atoms and i is the square root of  $-1$ . In this  $\pi$ -system, no circulation of  $\pi$ -electrons is expected along the circuits that share the  $p-q \pi$ -bond in common. BRE for the  $p-q\pi$ -bond is given as a destabilization energy of this hypothetical  $\pi$ -system. That is, BRE for a given  $\pi$ -bond represents the contribution of all circuits that share the bond to TRE. This quantity was originally defined to justify the isolated pentagon rule for fullerenes.**<sup>17</sup>** For porphyrinoids, SSE is equal to the BRE for any of the C–C bonds that link any pair of adjacent pyrrole rings.**5–7** vest college in the August 2010 Published on the college of this space of the space of the space of the Channel Conduction of the college of the c

Our theory of ring-current diamagnetism**19–22** is nothing other than an exact or analytical reformulation of Hückel–London theory.**23,24** It allows a decomposition of the ring current induced in a polycyclic  $\pi$ -system, G, exactly into individual circuit contributions. A current induced in each circuit may be termed a circuit current, the intensity of which is given in the form:**20–22**

$$
I_{i} = 18I_{0} \frac{S_{i}}{S_{0}} \prod_{m>n}^{r_{i}} k_{mn} \sum_{j}^{\text{occ}} \frac{P_{G-r_{i}}(X_{j})}{P_{G}'(X_{j})}
$$
(1)

where  $I_0$  is the intensity of a  $\pi$ -electron current induced in a benzene molecule;  $S_i$  and  $S_0$  are the areas of the *i*th circuit and the benzene ring, respectively;  $r_i$  is a set of conjugated atoms and  $\pi$ -bonds that constitute the *i*th circuit  $c_i$ ;  $k_{mn}$  is the Hückel parameter for the resonance integral between atoms *m* and *n*; *m* and *n* run over all  $\pi$ -bonds that belong to  $c_i$ ; G- $r_i$  is the subsystem of G, obtained by deleting  $r_i$  from G;  $P_G(X)$  and  $P_{G-r_i}(X)$  are the characteristic polynomials for G and  $G-r_i$ , respectively;  $X_i$  is the *j*th largest zero of  $P_G(X)$ ; a prime added to  $P_G(X)$  indicates the first derivative with respect to *X*; and *j* runs over all occupied  $\pi$ -molecular orbitals. If there are degenerate  $\pi$ -molecular orbitals, eqn (1) must be replaced by others.<sup>25,26</sup> Positive and negative  $I_i$ values represent diatropic and paratropic currents, respectively. A ring current distribution in an entire  $\pi$ -system can be obtained by superposing all circuit currents on G. It is exactly the same as that obtained by the original Hückel–London procedure.<sup>27,28</sup> Bifurcation of a ring current in a polycyclic  $\pi$ -system<sup>29</sup> can be associated reasonably with the circuit currents induced in it.

Hückel heteroatom parameters proposed by Van-Catledge<sup>30</sup> were employed. In addition, all nitrogen atoms coordinated to a metal ion were dealt with as imine  $(=N-)$  nitrogens. This assumption is fully consistent with the geometry of magnesium



**Fig. 2** Bond lengths (in  $\hat{A}$ ) in magnesium porphine (2,  $M = Mg$ ). Values in parentheses are Hückel  $\pi$ -bond orders.

porphine (**2**) and with the ring current distribution in it. Fig. 2 shows the molecular geometry of magnesium porphine calculated by Cyranski et al.<sup>31</sup> at the B3LYP/6-31G\* level of theory,<sup>32-34</sup> where all outer C–C bonds in pyrrole rings behave as formal double bonds.**<sup>35</sup>** As will be seen later, the ring current distribution we calculated for magnesium porphine is very similar to the current density map reported by Steiner and Fowler,**36,37** in which the current flowing along the inner cross is slightly stronger than the one flowing along the 20-membered outer periphery. Hückel parameters we employed for the nitrogen atoms reasonably reproduce these aspects of metalloporphyrins. There are two extra  $\pi$ -electrons in the inner ring, formally making a 16-membered aromatic dianion.

Realistic molecular geometries are necessary to evaluate the intensities of ring and circuit currents. Geometries used for **1–6** are those obtained by Jusélius and Sundholm<sup>38,39</sup> at the resolutionof-the-identity density-functional theory (RI-DFT) level**<sup>40</sup>** using the Becke–Perdew (B–P) parametrization**41-43** as implemented in TURBOMOLE.**<sup>44</sup>** We optimized molecular geometries of **7–9** at the B3LYP/6-31G\*\* level of theory**33,34** using a GAUSSIAN 03 program.**<sup>32</sup>** For the optimized geometries of **7–9**, see the ESI†. These geometries were used to calculate the areas enclosed by the circuits.

#### **3. Results and discussion**

We first explored macrocyclic aromaticity in the cores of natural porphyrinoids (**1–6**) and three expanded porphyrins (**7–9**), structural formulae of which are given in Fig. 2. Sapphyrin (**8**) is noted as the first expanded porphyrin.**45,46** Orangarin (**7**) and amethyrin (**9**) were prepared in 1995 by Sessler's group.**<sup>10</sup>** TREs and SSEs for these species are listed in Table 1, where  $\beta$  is the standard

**Table 1** TREs and SSEs for porphyrinoid species

Species	$TRE/ \beta $	$SSE/ \beta $
Porphine (1)	0.4322	0.0843
Metalloporphine (2)	0.4744	0.0795
Chlorin (3)	0.3955	0.0793
Metallochlorin (4)	0.4120	0.0761
Bacteriochlorin (5)	0.3171	0.0884
Metallobacteriochlorin (6)	0.3323	0.0812
Orangarin (7)	0.5656	$-0.0696$
Sapphyrin (8)	0.5904	0.0639
Amethyrin (9)	0.8020	$-0.0391$



**Fig. 3** Main macrocyclic conjugation pathways and BREs in nine porphyrinoids.

value for the resonance integral between two adjacent carbon 2p*<sup>z</sup>* orbitals. All these species are moderately aromatic with positive TREs. All species but **7** and **9** are slightly superaromatic with small positive SSEs, whereas **7** and **9** are slightly anti-superaromatic with small negative SSEs. It is obvious from these TREs and SSEs that macrocyclic conjugation never contributes significantly to the global aromaticity of a porphyrinoid  $\pi$ -system.<sup>5-7</sup> Macrocyclic annulene pathways proposed by previously<sup>1-6</sup> are shown in bold in Fig. 3. These annulene pathways are consistent with the signs of SSEs, in that aromatic and antiaromatic annulene pathways correspond to positive and negative SSEs, respectively.

BREs for all non-identical  $\pi$ -bonds in **1–9** are graphically summarized in Fig. 3. For porphyrinoids with positive SSEs (**1–6**, **8**), the main macrocyclic conjugation pathway can be traced by choosing a  $\pi$ -bond with a larger BRE at every bifurcation of the  $\pi$ -network.<sup>5,6</sup> Those in free-base porphyrins 1, 3, 5, and 8 are aromatic [4*n*+2]annulene pathways, which are all conjugated circuits in Randic's termonology.<sup>47</sup> For metal complexes 2, 4, and  $6$ ,  $\pi$ -bonds with larger BREs constitute the innermost, shortest possible, [16]annulene pathway along which  $18 \pi$ -electrons formally reside. All  $\pi$ -bonds located along these conjugation

pathways are intensified, with larger positive BREs than those located along the bypasses.

There are no such aromatic annulene pathways in orangarin (**7**) and amethyrin (**9**), because they have negative SSEs. Another type of macrocyclic pathways can instead be traced by choosing a  $\pi$ -bond with a smaller BRE at every bifurcation of the p-network.**5,6** Main macrocyclic pathways thus determined for **7** and **9** are antiaromatic [4*n*]annulene pathways, which are also conjugated circuits in Randic's sense.<sup>47</sup> These pathways will also be referred to as main macrocyclic conjugation pathways, because they are closely associated with macrocyclic antiaromaticity. All  $\pi$ -bonds located along these antiaromatic pathways are weakened, with smaller BREs than those located along the bypasses. Macrocyclic annulene pathways proposed by porphyrin chemists can thus be justified in terms of BREs.**1-7**

Next, let us examine the geometric patterns of macrocyclic circulation in porphyrinoids. Steiner and Fowler**8,9,36,37** carried out ring-current calculations on **1–3** and **7–9** with the 6-31G\*\* basis by means of coupled Hartree–Fock theory within the ipsocentric**48,49** CTOCD-DZ (continuous transformation of origin of current density diamagnetic zero) formulation.**50,51** They noted that all these porphyrinoids give clearly dominant macrocyclic currents;

intense diamagnetic currents are induced along the macrocycles of **1–3** and **8**, whereas the macrocycles of **7** and **9** sustain intense paramagnetic currents.**8,9** The occurrence of paramagnetic ring currents in **7** and **9** may be due partly to the absence of substituents. Sessler *et al.* reported that polysubstituted forms of **7** and **9** are  $20\pi$ - and  $24\pi$ -electron nonaromatic macrocycles, respectively.<sup>10</sup>

Main streams of Steiner–Fowler macrocyclic currents in porphine (**1**) **<sup>36</sup>** and sapphyrin (**8**) **<sup>8</sup>** follow the main macrocyclic conjugation pathways shown in Fig. 3. In metalloporphine (**2**) and chlorin (**3**), however, no distinct main stream is observable in the Steiner–Fowler current density maps;<sup>36,37</sup>  $\pi$ -currents of comparable intensity flow along the outer C–C and inner C–N bonds of one or more pyrrole rings. On the other hand, orangarin (**7**) and amethyrin (**9**) are predicted to sustain intense paramagnetic currents along the inner periphery, but not along the conventional annulene pathways.**8,9** The main streams of these paramagnetic currents necessarily pass through all nitrogen atoms including all amine nitrogens.

Global ring current patterns we calculated for **1–9** are shown as **1a**–**9a** in Fig. 4. Here, counterclockwise and clockwise circulations represent diatropic and paratropic ring currents, respectively. As far as  $1-3$  and  $7-9$  are concerned, our Hückel-London calculations reproduce well the global ring current patterns obtained by Steiner and Fowler.**8,9,36,37** Main streams of macrocyclic circulation in free bases **1**, **5**, and **8** follow the main conjugation pathways, whereas those in the other species do not always follow the main conjugation pathways. In  $2-4$  and  $6$ ,  $\pi$ -currents of comparable intensity flow along the outer CC and inner CN bonds of one or more pyrrole rings. Therefore, no distinct main streams are observable in these species. As noted by Steiner and Fowler,**8,9** the



**Fig. 4** Decomposition of ring currents induced in **1–9** (**1a**–**9a**) into macrocyclic (**1b**–**9b**) and local (**1c**–**9c**) circuit contributions. All current intensities are given in units of the benzene value.

main streams of macrocyclic ring currents in **7** and **9** runs along the inner periphery. These aspects of macrocyclic circulation do not conform to the annulene picture of porphyrinoids.

Although our ring current distributions for **7** and **9** indeed are similar to Steiner and Fowler's current density maps,**8,9** it seems that our simpler calculations somewhat overestimate the intensities of macrocyclic paramagnetic currents. This kind of drawback is possibly due to the neglect of bond-length alternation and/or deformation that might occur along the antiaromatic annulene pathways. Note that many porphyrinoid macrocycles with [4*n*]annulene pathways do not exhibit antiaromatic ring current effects and have been regarded as nonaromatic ones.**2–6** This experimental fact implies that macrocyclic antiaromaticity tends to be suppressed significantly by structural modifications. Steiner–Fowler current density maps for **7** and **98,9** can be reproduced formally by assigning a resonance integral of *ca.* 0.8*b* to all  $\pi$ -bonds that link adjacent pyrrolic rings. However, such a modification of the  $\pi$ -system does not affect the discussion below.

As has been seen above, at least for **7** and **9**, the main macrocyclic conjugation pathway is never the same as the main stream of macrocyclic circulation.**6,8,9** We then describe how to extract information on the main macrocyclic conjugation pathway from the ring current distribution. Note that local circulations within individual pyrrolic rings are superposed on macrocyclic circulation, both constituting the global current distribution.**<sup>7</sup>** Local circulations in the pyrrolic rings of **1–9** are always diamagnetic in nature. In the case of porphyrinoids with positive SSEs, a diamagnetic current is induced along the macrocycle; it is bifurcated when it runs across every pyrrole ring. Both diamagnetic macrocyclic and diamagnetic local currents flow in the same direction on the outer C–C bonds of the pyrrole rings. However, both currents totally or partially cancel each other out on the inner C–N bonds, because both currents flow in opposite directions there. Thus, local circulations may obscure the location of the main macrocyclic circulation pathway. main atesnes of macrocylic ring currents in 7 and 9 mm along<br>
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On the other hand, porphyrinoids with negative SSEs sustain a paramagnetic current along the macrocycle. The current is bifurcated when it runs across every pyrrole ring. Apart from this, a strong diamagnetic current is induced in every five-site circuit. In this case, both paramagnetic macrocyclic and diamagnetic local currents flow in opposite directions on the outer C–C bonds of the pyrrole rings and so totally or partially cancel each other out there. However, both currents flow in the same direction on the inner C–N bonds. As a result, the intensified current is observed on all inner C–N bonds. This must be the main reason why **7** and **9** appear to sustain strong paramagnetic circulations along the inner peripheries.

Therefore, if one wants to trace macrocyclic circulation only, the motion of  $\pi$ -electrons along the macrocycle only must be extracted from the global current distribution. Fortunately, our analytical theory of ring-current diamagnetism**20-22** allows us to decompose the ring current distribution exactly into individual circuit contributions. As shown in Fig. 4, the entire ring current in each porphyrinoid species can then be decomposed exactly into the contributions of macrocyclic and local five-site circuits. For example, the entire ring current induced in porphine (**1a**) can be partitioned into the contributions of macrocyclic (**1b**) and local five-site (**1c**) circuits; **1b** and **1c** consist of currents induced in 16 macrocyclic circuits (circuits **c–k** in Fig. 1) and 4 five-site ones

(circuits **a** and **b** in Fig. 1), respectively. It is **1b** that represents the net macrocyclic circulation induced in **1**. Comparison of **1a** with **1b** reveals that the main stream of macrocyclic circulation in **1a** remains unchanged in **1b**, which indicates that the main macrocyclic circulation pathway is the same as the main stream of circulation in the entire  $\pi$ -system. Both agree with the main macrocyclic conjugation pathway exactly. The same is true for two other free-base porphyrinoids **5** and **8**. Here, we tacitly assumed that the areas of all macrocyclic circuits are roughly of comparable magnitude, being much larger than those of local fivesite circuits. This possibly is a fairly reasonable assumption for all porphyrinoids studied.

The entire ring current in orangarin (**7a**) can be decomposed into the contributions of macrocyclic (**7b**) and local five-site (**7c**) circuits; **7b** and **7c** consist of currents induced in 32 macrocyclic and 5 five-site circuits, respectively. Likewise, the entire ring current in amethyrin (**9a**) can be decomposed into the contributions of 64 macrocyclic (**9b**) and 6 five-site (**9c**) circuits. That is, **7b** and **9b** represent net macrocyclic circulations induced in **7** and **9**, respectively. One sees that the main circulation pathways in **7b** and **9b** are different from the main streams of global circulation in **7a** and **9a**, respectively, but are identical with the main macrocyclic conjugation pathways in Fig. 1. As differences between the intensities of currents that flow on the outer C–C and inner C–N bonds are sizeable in **7b** and **9b**, there is no doubt that strong macrocyclic currents are induced along the conventional annulene pathways.

As stated above, **2–4** and **6** exhibit a current distribution pattern in which the  $\pi$ -currents of similar intensity flow along the outer C–C and inner C–N bonds in one or more pyrrole rings. These ring current patterns can be analyzed in the same manner. If we exclude local circulations in pyrrole rings from the ring current distribution as in **2b**–**4b** and **6b**, we can discern the same annulene pathways as predicted from the BREs. Thus, for all of **1–9**, authentic main macrocyclic conjugation pathways are clearly discernible in the macrocyclic current distributions (**1b**–**9b**) in which local circulations are missing. We can now safely say that, as far as porphyrinoid macrocycles are concerned, the main macrocyclic circulation pathway is identical with the main macrocyclic conjugation pathway. All main macrocyclic conjugation pathways pass through imine  $(=N-)$  nitrogens but avoid amine (–NH–) nitrogens.

Finally, one might have noted that some five-site or pyrrole circuits, such as those in **7c** and **9c**, sustain much larger diamagnetic currents than those in other porphyrinoids. It may be interesting to see that highly antiaromatic and aromatic circuits coexist in **7** and **9**. In such polycyclic  $\pi$ -systems, both the aromaticity of aromatic circuits and the antiaromaticity of antiaromatic circuits are often enhanced appreciably; paramagnetic and diamagnetic circuit currents induced in these circuits are necessarily intensified simultaneously. This is why all five-site circuits in **7** and **9** sustain large diamagnetic currents. Similar intensification of circuit currents was observed in many charged polycyclic benzenoid hydrocarbons.**<sup>52</sup>**

#### **4. Concluding remarks**

In 1989, Katritzky *et al.* pointed out that 'geometric-energetic aromaticity' is orthogonal to 'magnetic aromaticity'.**53,54** In fact,

aromatic stabilization is determined primarily by the connectivity of conjugated atoms, whereas  $\pi$ -circulation is a complicated function of the connectivity and geometry of conjugated atoms. It follows that, for porphyrinoids, a main route of macrocyclic circulation is not always the same as the main macrocyclic conjugation pathway. This is particularly true for antiaromatic macrocycles, such as orangarin (**7**) and amethyrin (**9**). The main streams of macrocyclic circulation in **7** and **9** pass through amine nitrogens,**8,9** although a main macrocyclic conjugation pathway in any free-base porphyrinoid species never passes through these atoms.**5,6** Such an apparent dichotomy was solved successfully by removing local circulations in the pyrrole rings from the global ring current distribution. As in other porphyrinoids, the main macrocyclic circulation pathways in **7** and **9** proved to follow their respective macrocyclic annulene pathways. At present, our theory of ring-current diamagnetism is the only tool for distinguishing macrocyclic currents from the local ones.**20-22** The theoretical approach proposed in this paper must be applicable to elucidating many other electronic and magnetic properties of macrocyclic  $\pi$ -systems consistently. IF consider and<br>bit is a completed on the property of the second of the consideration of the consideration<br>of consideration on the consideration of the consideration of the consideration of the consideration of the consid

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## **References**

- 1 E. Vogel, *Pure Appl. Chem.*, 1993, **65**, 143.
- 2 B. Franck and A. Nonn, *Angew. Chem., Int. Ed. Engl.*, 1995, **34**, 1795.
- 3 T. D. Lash, *Synlett*, 2000, 279.
- 4 J. L. Sessler and D. Seidel, *Angew. Chem., Int. Ed.*, 2003, **42**, 5134.
- 5 J. Aihara, *J. Phys. Chem. A*, 2008, **112**, 5305.
- 6 J. Aihara and H. Horibe, *Org. Biomol. Chem.*, 2009, **7**, 1939.
- 7 J. Aihara, E. Kimura and T. M. Krygowski, *Bull. Chem. Soc. Jpn.*, 2008, **81**, 826.
- 8 E. Steiner and P. W. Fowler, *Org. Biomol. Chem.*, 2004, **2**, 34.
- 9 E. Steiner and P. W. Fowler, *Org. Biomol. Chem.*, 2006, **4**, 2473.
- 10 J. L. Sessler, S. J. Weghorn, Y. Hiseada and V. Lynch, *Chem. Eur. J.*, 1995, **1**, 56.
- 11 J. Aihara, *J. Am. Chem. Soc.*, 1976, **98**, 2750.
- 12 I. Gutman, M. Milun and N. Trinajstic,´ *J. Am. Chem. Soc.*, 1977, **99**, 1692.
- 13 A. Graovac, I. Gutman, N. Trinajstić and T. Zivković, Theor. Chim. *Acta*, 1972, **26**, 67.
- 14 V. I. Minkin, M. N. Glukhovtsev and B. Ya. Simkin, *Aromaticity and Antiaromaticity: Electronic and Structural Aspects*, Wiley-Interscience, New York, 1994, Ch. 3.
- 15 J. Aihara, *J. Am. Chem. Soc.*, 1992, **114**, 865.
- 16 J. Aihara, *J. Chem. Soc., Faraday Trans.*, 1995, **91**, 237.
- 17 J. Aihara, *J. Am. Chem. Soc.*, 1995, **117**, 4130.
- 18 J. Aihara, *J. Chem. Soc., Perkin Trans. 2*, 1996, 2185.
- 19 M. Makino and J. Aihara, *Phys. Chem. Chem. Phys.*, 2008, **10**, 591 (Electronic Supplementary Information).
- 20 J. Aihara, *J. Am. Chem. Soc.*, 1985, **107**, 298.
- 21 J. Aihara, *Bull. Chem. Soc. Jpn.*, 2004, **77**, 651.
- 22 J. Aihara, *J. Am. Chem. Soc.*, 2006, **128**, 2873.
- 23 F. London, *J. Phys.*, 1937, **8**, 397.
- 24 B. Pullman and A. Pullman, *Les Theories Electroniques de la Chimie ´ Organique*, Masson et Cie, Paris, 1952, Ch. IX.
- 25 J. Aihara, *J. Am. Chem. Soc.*, 1979, **101**, 5913.
- 26 J. Aihara and T. Horikawa, *Chem. Phys. Lett.*, 1983, **95**, 561.
- 27 J. A. Pople, *Mol. Phys.*, 1958, **1**, 175.
- 28 R. McWeeny, *Mol. Phys.*, 1958, **1**, 311. 29 E. Steiner and P. W. Fowler, *Org. Biomol. Chem.*, 2003, **1**, 1785.
- 30 F. A. Van-Catledge, *J. Org. Chem.*, 1980, **45**, 4801.
- 31 M. C. Cyrański, T. M. Krygowski, M. Wisiorowski, M. N. J. R. van Eikema Hommes and P. v. R. Schleyer, *Angew. Chem., Int. Ed.*, 1998, **37**, 177.
- 32 M. J. Frisch, G. W. Trucks, H. B. Schlegel, G. E. Scuseria, M. A. Robb, J. R. Cheeseman, J. A. J. Montgomery, T. Vreven, K. N. Kudin, J. C. Burant, J. M. Millam, S. S. Iyengar, J. Tomasi, V. Barone, B. Mennucci, M. Cossi, G. Scalmani, N. Rega, G. A. Petersson, H. Nakatsuji, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima , Y. Honda, O. Kitao, H. Nakai, M. Klene, X. Li, J. E. Knox, H. P. Hratchian, J. B. Cross, C. Adamo, J. Jaramillo, R. Gomperts, R. E. Stratmann, O. Yazyev, A. J. Austin, R. Cammi, C. Pomelli, J. W. Ochterski, P. Y. Ayala, K. Morokuma, G. A. Voth, P. Salvador, J. J. Dannenberg, V. G. Zakrzewski, S. Dapprich, A. D. Daniels, M. C. Strain, O. Farkas, D. K. Malick, A. D. Rabuck, K. Raghavachari, J. B. Foresman, J. V. Ortiz, Q. Cui, A. G. Baboul, S. Clifford, J. Cioslowski, B. B. Stefanov, G. Liu, A. Liashenko, P. Piskorz, I. Komaromi, R. L. Martin, D. J. Fox, T. Keith, M. A. Al-Laham, C. Y. Peng, A. Nanayakkara, M. Challacombe, P. M. W. Gill, B. Johnson, W. Chen, M. W. Wong, C. Gonzalez, and J. A. Pople, *Gaussian 03*, Revision B.02, Gaussian, Inc., Pittsburgh, PA, 2003.
- 33 C. Lee, W. Yang and R. G. Parr, *Phys. Rev. B*, 1988, **37**, 785.
- 34 A. D. Becke, *J. Chem. Phys.*, 1993, **98**, 5648.
- 35 E. B. Fleischer, *Acc. Chem. Res.*, 1970, **3**, 105.
- 36 E. Steiner and P. W. Fowler, *ChemPhysChem*, 2002, **3**, 114.
- 37 E. Steiner, A. Soncini and P. W. Fowler, *Org. Biomol. Chem.*, 2005, **3**, 4053.
- 38 J. Juselius and D. Sundholm, ´ *Phys. Chem. Chem. Phys.*, 2000, **2**, 2145.
- 39 J. Jusélius and D. Sundholm, *J. Org. Chem.*, 2000, 65, 5233.
- 40 K. Eichkorn, O. Treutler, H. Öhm, M. Häser and R. Ahlrichs, Chem. *Phys. Lett.*, 1995, **240**, 283.
- 41 S. H. Vosko, L. Wilk and M. Nusair, *Can. J. Phys.*, 1980, **58**, 1200.
- 42 J. P. Perdew, *Phys. Rev. B*, 1986, **33**, 8822.
- 43 A. D. Becke, *Phys. Rev. B*, 1988, **38**, 3098.
- 44 R. Ahlrichs, M. Bär, M. Häser, H. Horn and C. Kölmel, *Chem. Phys. Lett.*, 1989, **162**, 165.
- 45 M. J. Broadhurst, R. Grigg and A. W. Johnson, *J. Chem. Soc., Perkin Trans. 1*, 1972, 2111.
- 46 V. J. Bauer, D. L. J. Clive, D. Dolphin, J. B. Paine, F. L. Harris, M. M. King, J. Loder, S.-W. C. Wang and R. B. Woodward, *J. Am. Chem. Soc.*, 1983, **105**, 6429.
- 47 M. Randic,´ *J. Am. Chem. Soc.*, 1977, **99**, 444.
- 48 E. Steiner and P. W. Fowler, *Chem. Commun.*, 2001, 2220.
- 49 E. Steiner and P. W. Fowler, *J. Phys. Chem.*, 2001, **105**, 9553.
- 50 T. A. Keith and R. F. W. Bader, *Chem. Phys. Lett.*, 1993, **210**, 223.
- 51 S. Coriani, P. Lazzeretti, R. Malagoli and R. Zanasi, *Theor. Chim. Acta*, 1994, **89**, 181.
- 52 T. Ishida, H. Kanno and J. Aihara, *Pol. J. Chem.*, 2007, **81**, 699.
- 53 A. R. Katritzky, P. Barczynski, G. Musumarra, D. Pisano and M. Szafran, *J. Am. Chem. Soc.*, 1989, **111**, 7.
- 54 J. Aihara, *Bull. Chem. Soc. Jpn.*, 2008, **81**, 241.